

Mole Potpourri Worksheet Answers

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Mole mass practice worksheet answers

The Mega Mole Worksheet #1-10 Convert to Moles 12.04 x 10²³ atoms He. 3.01 x 10²³ atoms Cu. 3.612 x 10²³ atoms Fe. 100 atoms Ar. 1 atom S. 24 grams S. 59.3 grams Sn. 98.9 grams Na. 5000 grams K. 0.000500 grams Ne #11-15 Convert to mass in grams. 10.0 moles Na. 2.20 moles Sn. 5.00 moles Ag. 0.0003000 moles Au. 1.00 x 10²³ moles B #16-20 Convert ...

The Mega Mole Worksheet

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Mole Calculation Worksheet - Answer Key What are the molecular weights of the following compounds? 1) NaOH $22.99 + 16.00 + 1.01 = 40.00$ grams/mol 2) H_3PO_4 $3(1.01) + 30.97 + 4(16.00) = 98.00$ grams

Mole Calculation Worksheet - Brookside High School

Related Topics: More Lessons for Chemistry Math Worksheets Mole, Mass & Avogadro Constant. An amount of substance containing 6.02×10^{23} particles is called a mole (often abbreviated to mol).; 6.02×10^{23} is called the Avogadro Constant or Avogadro's Number.; The following diagram shows how to convert between Mass, Mole and Number of particles.

Mole, Avogadro Constant & Molar Mass (solutions, examples ...

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